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Chemistry Examples, **Chapter 12**
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PLAY. kinetic-molecular theory.
the behavior of matter in terms of
particles in motion. elastic
collision. collision in which no
kinetic energy is lost.
temperature. measure of the
average kinetic energy of the
particles in a sample of matter.

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MANUAL 30. Compare and
contrast sublimation and

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evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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motion 2. a. small b. forces c.
random d. elastic; kinetic 3. KE
 $\frac{1}{2} mv^2$ 4. Temperature 5. true 6.

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true 7. false 8. true 9. true 10.
false 11. true 12. false 13. a 14. a
15. d 16. d 17. b 18. b 19. b 20.
barometer 21. Evangelista
Torricelli

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collision. Kinetic energy formula.
Temperature. Describes the
behavior of matter in terms of
particles in moti.... No kinetic
energy is lost. $KE = \frac{1}{2}mv^2$.
Measure of the average kinetic
energy of the particles in a sa....

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